Dispiro[2,2,2,2]deca-4,9-diene: Preparation and Reactions

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Summary Dispiro[2,2,2,2]deca-4,9-diene (4) is prepared and its chemical and physical properties are examined; the reaction with dimethyl *trans,trans*-muconate affords the cycloadduct (5).

DE MEIJERE has recently reported the preparation of dispiro[2,0,2,4]deca-7,9-diene,¹ which has prompted us to report our results on the preparation and reactions of dispiro-[2,2,2,2]deca-4,9-diene (4), where the cyclopropane rings and the double bonds are aligned in the bisected conformation.²



2,5-Dimethylenecyclohexane-1,4-diol $(1)^3$ was cyclopropanated with methylene iodide and a Zn-Cu couple⁴ to dispiro[2,2,2,2]deca-4,9-diol (2) (50%), m.p. 219-220°,†

† Satisfactory analytical results were obtained for all new compounds.

which was subsequently oxidized with Jones reagent⁵ to the diketone (3) (69%), m.p. 81—82°. Treatment of the ditosylhydrazone of (3) with excess of BuⁿLi⁶ afforded (4) as highly volatile, colourless plates melting at 121—122° (64%), m/e 132 (M^+); τ (CCl₄): 5·20 (s, 4H) and 9·33 (s, 8H). Alternatively, pyrolysis of the bisxanthate of (2) at 300° under reduced pressure also afforded (4). The compounds obtained from the two independent preparations were identical in all respects.



The u.v. maximum of (4), λ_{max} (n-hexane) 222 nm (log ϵ 4·30), is at a longer wavelength than that of tetracyclopropylethylene, λ_{max} (n-hexane) 219 nm (log ϵ 4·08).⁷ However, the possibility of cyclic delocalisation of electrons in the six-membered ring in (4) is ruled out by the n.m.r. spectrum. The signal due to the olefinic protons of (4) (τ 5·20) appears at a higher field than that for the 4-H in spiro[2,5]oct-4-ene (τ 5·10).⁸ Thus, no diamagnetic ring current is observed in (4).

Compound (4) was moderately thermally stable; its halflife at 160° was *ca.* 1 h. The major product of thermal decomposition in diglyme was p-diethylbenzene (75%).

Compound (4) reacted smoothly with dimethyl trans,trans-muconate in benzene at 160° to give a colourless crystalline cycloadduct (5) (54%), m.p. 144—145°; m/e 302 (M^+) .[‡] Comparison of the n.m.r. and u.v. spectra with those of reported [8]paracyclophane derivatives⁹ confirm the paracyclophane structure of (5).

 $[\sigma 2 + \pi 2 + \sigma 2 + \pi 4]$ process, although it is not known whether the cycloaddition proceeds in a concerted manner or stepwise.

This cycloaddition can be considered formally as a

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Though the stereochemistry of the product has not been fully established, the n.m.r. is best accounted for by the configuration in which the methoxycarbonyl groups are trans with respect to the 12-C ring, and the double bond on the bridge is also trans.

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